

**IN THE SPECIFICATION:**

**Page 13, seventh full paragraph:**

a1  
The "solution containing silver ions" used in the present invention means a solution containing silver ions in ~~wafer~~ water or a mixture of ~~wafer~~ water and an organic solvent. As the ion source of the solution containing silver ions, a ~~wafer-soluble~~ water-soluble silver salt can be used. As the ~~wafer-soluble~~ water-soluble silver salt, silver nitrate is preferably used. The concentration of the silver salt in the solution containing silver ions is preferably 0.03-6.5 mol/L, more preferably 0.1-5 mol/L. The pH of the solution containing silver ions is preferably 2-6, more preferably 3.5-6. For the pH adjustment, acids and alkalines usually used for pH adjustment may be added to the solution containing silver ions. The solution containing silver ions preferably has a Reynolds number in the range of 500-20,000, when the solution containing silver ions is supplied.

**Page 16, third full paragraph:**

a2  
The "reaction field solution" used for the present invention consists of ~~wafer~~ water or a mixed aqueous solution of ~~wafer~~ water and an organic solvent. The reaction field solution may contain a dispersing agent and so forth, and it may be a recycled portion of the organic acid silver salt solution obtained in the reaction.